

The First Total Synthesis of the Novel β -Carboline Alkaloid Oxopropaline G

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Abstract: The first total synthesis of the new type of cytocydal β -carboline alkaloid oxopropaline G was achieved in 12 steps. © 1998 Elsevier Science Ltd. All rights reserved.

Oxopropalines (1) isolated from *Streptomyces* sp. G324^{1,2} producing lavendamycin,³ are novel β -carboline alkaloids. These compounds are structurally composed of five constituent compounds, A, B, D (1b), E and G (1a), were elucidated by physico-chemical studies by Abe and co-workers.² This new type of β -carboline alkaloids, possessing an acyl group and a methyl group at the 1- and 4-positions, respectively, exhibit the cytocydal activity.¹

We are currently developing the synthesis of biologically active condensed-heteroaromatic compounds, including natural products by the thermal electrocyclic reaction of either hexatriene⁴ or

monoazahexatriene systems⁵ incorporating one double bond of aromatic or heteroaromatic portion. We here describe the first total synthesis of oxopropaline G (1a) by application of this pyrido-annelation. The present methodology is based on the thermal electrocyclic reaction of 1-azahexatriene system (5) involving the indole 2,3-bond to prepare a new 4-methyl- β -carboline ring (4). 1-Methoxycarbonyl-4-methyl- β -carboline (2) derived from 4 was envisaged as a key compound for the total synthesis of oxopropaline G (1a) and D (1b) (Scheme 1).

Scheme 2

The required β -carboline (4) was prepared in four steps starting from 2-formyl-3-iodoindole (6)^{sc} (Scheme 2). The cross-coupling reaction between 6 and isopropenyl tributyltin^{6b} in the presence of $PdCl_2(PPh_3)_2$ in DMF gave the isopropenylindole (7) (93%),⁷ which on *N*-protection with chloromethyl methyl ether (MOMCl) provided *N*-MOM-indole (8) (99%). Subsequent treatment of 8 with hydroxylamine produced the oxime (5) as the 1-azahexatriene system, which was subjected to the thermal electrocyclic reaction in o-dichlorobenzene (190 °C, 1 hr) to yield the 4-methyl- β -carboline (4) (81% from 8).

The key compound (2) was synthesized from β -carboline (4) in four steps. Treatment of 4 with m-chloroperbenzoic acid (mCPBA) followed by heating in acetic anhydride (Ac_2O) yielded the 1-hydroxy- β -carboline (10) (94% from 5), which was treated with trifluoromethnesulfonic anhydride (Tf_2O) to obtain the triflate (3) (80%). The triflate (3) was converted to the desired 1-methoxycarbonyl-4-methyl- β -carboline (2) with the three component cross-coupling reaction (a.8) [triflate (3), carbon monooxide (1 atm), methanol, triethylamine, $Pd(OAc)_2$, and 1,1'-bis(diphenylphosphino)-ferrocene (dppf) in DMF] in 74% yield.

At the final stage, oxopropaline G (1a) was synthesized from the key compound (2) in four steps. Nucleophilic addition reaction to 2 with the acetate carbanion⁹ [prepared from ethyl acetate with lithium hexamethyldisilazide (LHMDS)] produced the β -keto ester (*N*-MOM, 11) (99%), which was deprotected with trifluoromethanesulfonic acid in the presence of methanol and trimetyl orthoformate in nitromethane¹⁰ to give the keto ester (12) (71%). Reduction of 12 with diborane in THF followed by selective oxidation of the 1,3-diol (13) with activated manganase dioxide provided oxopropaline G (1a) (40% from 12). All spectral data of synthetic oxopropaline G (1a)¹¹ was in good agreement with that reported for the natural product.²

In conclusion, the first total synthesis of a new type of cytocydal β -carboline alkaloid oxopropaline G (1a) was completed in a twelve-step sequence (11.7% overall yield from 6) by thermal electrocyclic reaction of a 1-azahexatriene system involving the indole 2,3-bond for the construction of the β -carboline framework (4) followed by formation of the acyl group at the 1-position *via* the key compound, 1-methoxycarbonyl-4-methyl- β -carboline (2).

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References and Notes

- 1. Abe, N.; Nakakita, Y.; Nakamura, T.; Enoki, N.; Uchida, H.; Takeo, S.; Munekata, M. *J. Antibiotics* **1993**, *46*, 1672.
- 2. Abe, N.; Enoki, N.; Nakakita, Y.; Uchida, H.; Nakamura, T.; Munekata, M. J. Antibiotics 1993, 46, 1678.

- 3. Balitz, D.M.; Bush, J.A.; Bradner, W.T.; Doyle, T.W.; O'Herron, F.A.; Nettleton, D.E. J. Antibiotics 1982, 35, 256.
- (a) Okamura, W.H.; de Lera, A.R. In Comprehensive Organic Synthesis; Trost, B.M.; Fleming, I.;
 Paquette, L.A., Eds.; Pergamon Press: New York, 1991; Vol. 5, pp. 697-750. (b) Hibino, S.;
 Sugino, E. In Advances in Nitrogen Heterocycles; Moody, C.J.; Ed.; JAI Press: Greenwich, CT, 1995; Vol. 1, pp. 205-227.
- (a) Choshi, T.; Sada, T.; Fujimoto, H.; Nagayama, C.; Sugino, E.; Hibino, S. *Tetrahedron Lett.* 1996, 37, 2593. (b) Choshi, T.; Fujimoto, H.; Sugino, E.; Hibino, S. *Heterocycles* 1996, 43, 1847. (c) Choshi, T.; Sada, T.; Fujimoto, H.; Nagayama, C.; Sugino, E.; Hibino, S. *J. Org. Chem.* 1997, 62, 2535.
- (a) Choshi, T.; Yamada, S.; Sugino, E.; Kuwada, T.; Hibino, S. J. Org. Chem. 1995, 60, 5899.
 (b) Yoshioka, H.; Choshi, T.; Sugino, E.; Hibino, S. Heterocycles 1995, 41, 161. (c) Yoshioka, H.; Matsuya, Y.; Choshi, T.; Sugino, E.; Hibino, S. Chem. Pharm. Bull. 1996, 44, 709 and related references cited therein.
- 7. All new compounds provided satisfactory spectral and analytical data.
- 8. Cacci, S.; Ciatini, P.G.; Morera, E.; Ortar, G. Tetrahedron Lett. 1986, 27, 3931.
- 9. The Claisen reaction was carried out under the conditions cited in the abstract: Suzuki, H.; Ebihara, Y.; Yokoyama, Y.; Murakami, Y. The 27th Congress of Heterocyclic Chemistry, October 11-13, 1996 (Morioka, Japan), abstract (3A-03) p.288.
- 10. This reaction conditions reported by the following literature was utilized for the ketalization of ketone group of 11. However, the deprotection of *N*-MOM group of 11 occurred instead of the ketalization: Thurkauf, A.; Jacobson, A.E.; Rice, K.C. *Synthesis* 1988, 233.
- 11. Oxopropaline G: mp 155-157 °C (CHCl₃); Ir (KBr) 3518, 3320, 1663 cm⁻¹; ¹H-nmr (500 MHz, MeOH-d₄) 2.90 (3H, s), 3.54 (2H, t, *J*=6.4 Hz), 4.07 (2H, t, *J*=6.4 Hz), 7.33 (1H, dd, *J*=7.9, 7.1 Hz), 7.58 (1H, dd, *J*=8.2, 7.1 Hz), 7.71 (1H, d, *J*=8.2 Hz), 8.23 (1H, d, *J*=7.9 Hz), 8.24 (1H, s); ¹³C-nmr (125 MHz, MeOH-d₄) 17.9, 41.8, 58.8, 113.3, 121.7, 122.1, 124.5, 129.6, 131.1, 134.0, 135.2, 135.9, 139.7, 143.2, 203.1. Ms (CI) *m/z* 255 (M⁺+1).